## 2014 Consumer Confidence Report

Report Date: Water System Name: Goldside/Hillview May 1, 2015

We test the drinking water quality for many constituents as required by state and federal regulations. This report shows the results of our monitoring for the period of January 1 - December 31, 2014 and may include earlier monitoring data.

Este informe contiene información muy importante sobre su agua potable. Tradúzcalo ó hable con alquien que lo entienda bien.

Type of water source(s) in use: Hard rock wells which draw from underground fractures.

Name & general location of source(s): Hillview Estates – Well #1, Goldside – Wells #2, #4, #6, and #7. River Creek Wells #1, and #2, and Miami Creek #1.

Drinking Water Source Assessment Information: A source water assessment was conducted for the active water supply wells of the Hillview Water Company, Inc. - Goldside by the Department of Health Services on August 20, 2002. The sources considered most vulnerable to the following activities not associated with any detected contaminants: Wells – Water supply, septic system – low density, sewer collection systems. A copy of the complete assessment may be viewed at the Hillview Water Company, Inc. 40312 Greenwood Way. Oakhurst, CA 93644. You may request a summary of the assessment be sent to you by contacting Ralph Fairfield 559.683.4322, P.O. 2269, Oakhurst, CA 93644.

Time and place of regularly scheduled board meetings for public participation: Hillview Water Company, Inc. does not hold regularly scheduled meetings. The public is allowed to participate in all CPUC proceedings.

For more information, contact: Hillview Water Company, Inc. Phone: 559.683.4322

## TERMS USED IN THIS REPORT

Maximum Contaminant Level (MCL): The highest Primary Drinking Water Standards (PDWS): MCLs and MRDLs MCLGs) as is economically and technologically feasible. Secondary MCLs are set to protect the odor, taste, and appearance of drinking water.

Maximum Contaminant Level Goal (MCLG): The MCL levels. level of a contaminant in drinking water below which Treatment Technique (TT): A required process intended to are set by the U.S. Environmental Protection Agency (USEPA).

Public Health Goal (PHG): The level of contaminant in drinking water below which there is no California Environmental Protection Agency.

Maximum Residual Disinfectant Level (MRDL) The highest level of a disinfectant allowed in drinking ND: not detectable at testing limit water. There is convincing evidence that addition of a disinfectant is necessary for control of microbial contaminants.

Maximum Residual Disinfectant Level Goal ppt: parts per trillion or nanograms per liter (ng/L) (MRDLG): The level of a drinking water disinfectant below which there is no known or expected risk to ppq: parts per quadrillion or picogram per liter (pg/L) health. MRDLGs do not reflect the benefits of the use pci/L: picocuries per liter (a measure of radiation) of disinfectants to control microbial contaminants.

level of a contaminant that is allowed in drinking water. for contaminants that affect health along with their monitoring and Primary MCLs are set as close to the PHGs (or reporting requirements, and water treatment requirements.

> Secondary Drinking Water Standards (SDWS): contaminants that affect taste, odor, or appearance of the drinking water. Contaminants with SDWSs do not affect the health at the

> reduce the level of a contaminant in drinking water.

Regulatory Action Level (AL): The concentration of a contaminant which, if exceeded, triggers treatment or other requirements that a water system must follow.

known or expected risk to health. PHGs are set by the Variances and Exemptions: State Board permission to exceed an MCL or not comply with a treatment technique under certain conditions.

**ppm**: parts per million or milligrams per liter (mg/L)

ppb: parts per billion or micrograms per liter (µg/L)

The sources of drinking water (both tap water and bottled water) include rivers, lakes, streams, ponds, reservoirs, springs, and wells. As water travels over the surface of the land or through the ground, it dissolves naturally-occurring minerals and, in some cases, radioactive material, and can pick up substances resulting from the presence of animals or from human activity.

## Contaminants that may be present in source water include:

- Microbial contaminants, such as viruses and bacteria, that may come from sewage treatment plants, septic systems, agricultural livestock operations, and wildlife.
- [ Inorganic contaminants, such as salts and metals, that can be naturally-occurring or result from urban stormwater runoff, industrial or domestic wastewater discharges, oil and gas production, mining, or farming.
- [ Pesticides and herbicides, that may come from a variety of sources such as agriculture, urban stormwater runoff, and residential uses.
- Organic chemical contaminants, including synthetic and volatile organic chemicals, that are by-products of industrial processes and petroleum production, and can also come from gas stations, urban stormwater runoff, agricultural application, and septic systems.
- [ Radioactive contaminants, that can be naturally-occurring or be the result of oil and gas production and mining activities.

In order to ensure that tap water is safe to drink, the USEPA and the State Water Resources Control Board (State Board) prescribe regulations that limit the amount of certain contaminants in water provided by public water systems. State Board regulations also establish limits for contaminants in bottled water that provide the same protection for public health.

Tables 1, 2, 3, 4, and 5 list all of the drinking water contaminants that were detected during the most recent sampling for the constituent. The presence of these contaminants in the water does not necessarily indicate that the water poses a health risk. The State Board allows us to monitor for certain contaminants less than once per year because the concentrations of these contaminants do not change frequently. Some of the data, though representative of the water quality, are more than one year old.

TABLE 1 – SAMPLING RESULTS SHOWING THE DETECTION OF COLIFORM BACTERIA						
Microbiological Contaminants (complete if bacteria detected)	Highest No. of Detections	No. of months in violation	MCL	MCLG	Typical Source of Bacteria	
Total Coliform Bacteria	(In a mo.)	More than 1 sample in a month with a detection.		0	Naturally present in the environment.	
Fecal Coliform or E. coli	(In the year) 0	0	A routine sample and a repeat sample		Human and animal fecal waste.	

TABLE 2 – SAMPLING RESULTS SHOWING THE DETECTION OF LEAD AND COPPER							
Lead and Copper (complete if lead or copper detected in the last sample set)	Sample Date	No. of samples collected	90 <sup>th</sup> percentile level detected	No. sites exceeding AL	AL	PHG	Typical Source of Contaminant
Lead (ppb)	9/11,13/13	10	ND	0	15	0.2	Internal corrosion of household water plumbing systems; discharges from industrial manufacturers; erosion of natural deposits.
Copper (ppm)	9/11,13/13	10	ND	0	1.3	0.3	Internal corrosion of household plumbing systems; erosion of natural deposits; leaching from wood preservatives.

TABLE 3 – SAMPLING RESULTS FOR SODIUM AND HARDNESS							
Chemical or Constituent (and reporting units)	Sample Date	Level Detected	Range of Detections	MCL	PHG (MCLG)	Typical Source of Contaminant	
Sodium (ppm)	6/24/14	47.8	10 – 170	None	None	Salt present in the water and is generally naturally occurring.	
Hardness (ppm)	6/24/14	164	57 – 300	None	None	Sum of polyvalent cations present in the water, generally magnesium and calcium, and are usually naturally occurring.	

<sup>\*</sup>Any violation of an MCL or AL is asterisked. Additional information regarding the violation is provided later in this report.

TABLE 4 – DETECTION OF CONTAMINANTS WITH A PRIMARY DRINKING WATER STANDARD							
Chemical or Constituent (and reporting units)	Sample Date	Level Detected	Range of Detections	MCL [MRDL]	PHG (MCLG) [MRDLG]	Typical Source of Contaminant	
Arsenic – ppb	3/3/14	ND	ND	10	0.004	Erosion of natural deposits; runoff from orchards; glass and electronics production wastes.	
Chlorine – ppm	January - December	1.14	0.68 – 1.49	[4]	[4]	Drinking water disinfectant added for treatment.	
Fluoride – ppm	3/3/14	0.08	ND – 0.38	2	1	Erosion of natural deposits; water additive which promotes strong teeth; discharge from fertilizer and aluminum factories.	
Gross Alpha Activity – pCi/L	3/3/14	13.12	ND – 22.27	15	(0)	Erosion of natural deposits.	
HAA5 (Haloacetic Acids) – ppb	7/23/14	<2	<2	60	NA	By-product of drinking water disinfection.	
Nitrate (as nitrate, NO <sub>3</sub> ) – ppm	3/3/14	8.29	ND – 21	45	45	Runoff and leaching from fertilizer use; leaching from septic tanks and sewage; erosion of natural deposits.	
Selenium – ppb	9/2009	3.2	ND – 3.2	50	30	Discharge from petroleum, glass, and metal refineries; erosion of natural deposits; discharge from mines and chemical manufacturers; runoff from livestock lots (feed additive).	
Toluene – ppb	9/2012	0.55	ND - 0.55	150	150	Discharge from petroleum and chemical factories; underground gas tank leaks.	
TTHMs (Total Trihalomethanes) – ppb	6/30/14	2.0	2.0	80	NA	By-product of drinking water disinfection.	
Uranium – pCi/L	3/3/14	4.9	4.9	20	0.43	Erosion of natural deposits.	
1,2 Dichlorobenzene – ppb	5/2012	5.9	ND – 5.9	600	600	Discharge from factories, dry cleaners, and auto shops (metal degreaser).	

TABLE 5 – DETECTION OF CONTAMINANTS WITH A <u>SECONDARY</u> DRINKING WATER STANDARD								
Chemical or Constituent (and reporting units)	Sample Date	Level Detected	Range of Detections	MCL	PHG (MCLG)	Typical Source of Contaminant		
Chloride – ppm	3/10/14	129.78	3.9 – 410	500	NA	Runoff/leaching from natural deposits; seawater influence.		
Chlorine – ppm	9/2012	430	340 – 430	500	NA	Runoff/leaching from natural deposits; industrial wastes.		
Color – Units	3/10/14	2	<5 – 10	15	NA	Naturally-occurring organic materials.		
Copper – ppm	3/10/14	ND	ND	1	NA	Internal corrosion of household plumbing systems; erosion of natural deposits; leaching from wood preservatives.		
Iron – ppb	3/10/14	190	ND - 900	300	NA	Leaching from natural deposits.		
Manganese – ppb	10/23/14	27.57	ND – 83	50	NA	Leaching from natural deposits.		
Specific Conductance µS/cm	3/18/14	704	165 – 1500	1600	NA	Substances that form ions when in water; seawater influence.		
Sulfate – ppm	3/18/14	9	<2 – 26	500	NA	Runoff/leaching from natural deposits; industrial wastes.		
Total Dissolved Solids (TDS) – ppm	3/18/14	466	130 – 1100	1000	NA	Runoff/leaching from natural deposits.		
Turbidity – Units	3/18/14	<1	ND – 4.7	5	NA	Soil runoff.		
Zinc – ppm	3/18/14	ND	ND	5	NA	Runoff/leaching from natural deposits; industrial wastes.		

## Additional General Information on Drinking Water

Drinking water, including bottled water, may reasonably be expected to contain at least small amounts of some contaminants. The presence of contaminants does not necessarily indicate that the water poses a health risk. More information about contaminants and potential health effects can be obtained by calling the USEPA's Safe Drinking Water Hotline (1-800-426-4791).

Some people may be more vulnerable to contaminants in drinking water than the general population. Immuno-compromised persons such as persons with cancer undergoing chemotherapy, persons who have undergone organ transplants, people with HIV/AIDS or other immune system disorders, some elderly, and infants can be particularly at risk from infections. These people should seek advice about drinking water from their health care providers. USEPA/Centers for Disease Control (CDC) guidelines on appropriate means to lessen the risk of infection by *Cryptosporidium* and other microbial contaminants are available from the Safe Drinking Water Hotline (1-800-426-4791).

Lead-Specific Language for Community Water Systems: If present, elevated levels of lead can cause serious health problems, especially for pregnant women and young children. Lead in drinking water is primarily from materials and components associated with service lines and home plumbing. Hillview Water Company, Inc. is responsible for providing high quality drinking water, but cannot control the variety of materials used in plumbing components. When your water has been sitting for several hours, you can minimize the potential for lead exposure by flushing your tap for 30 seconds to 2 minutes before using water for drinking or cooking. If you are concerned about lead in your water, you may wish to have your water tested. Information on lead in drinking water, testing methods, and steps you can take to minimize exposure is available from the Safe Drinking Water Hotline or at <a href="http://www.epa.gov/safewater/lead">http://www.epa.gov/safewater/lead</a>.